

lab 4 intermolecular forces pdf

Laboratory: Intermolecular Forces (IMF) Report Requirement: Answer all of the questions/do all the computations requested in italics. Questions not in italics do NOT need to be answered. You do NOT have to write a formal lab report.

Laboratory: Intermolecular Forces (IMF) - Parkland College

Intermolecular forces (IMF) ... compounds can burn we focus in this lab on the hydrocarbons. ... 4 (water soluble) + Saturated hydrocarbons are unreactive ADVISING HANDBOOK - North Carolina State University

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In today's lab we will examine the attractive forces that hold molecules together and the disruptive forces that break them apart. The forces between molecules that hold ... 4. List the following intermolecular forces in order of increasing strength:

Lab #20 Intermolecular Forces - Weebly

intermolecular forces and the effect of intermolecular forces on the surface tension of water. Students will use the information gathered during their research to predict how different factors affect the strength of the intermolecular forces and the resulting surface tension of water, and then design and carry out a lab to test their hypothesis.

The Effect of Intermolecular Forces on Surface Tension

CHEM 1112 General Chemistry II Laboratory Page 1 Laboratory 4. Intermolecular Forces 1 Team Members (print and sign name by role) Analyst 1: _____ (oldest) Contributes by: actively participating as a learner, keeping team on task, assuring that all team members participate and understand, making certain each team member is performing within their role.

Lab4 - CHEM 1112 General Chemistry II Laboratory

Intermolecular forces- forces of attraction and repulsion between molecules that hold molecules, ions, and atoms together. Intramolecular - forces of chemical bonds within a molecule Boiling Point and Electronegativity Boiling Point - the temperature at which the liquid form of a compound vaporizes into a gas.

INTERMOLECULAR FORCES - UCLA

Evaporation and Intermolecular Attractions In this experiment, Temperature Probes are placed in various liquids. ... like viscosity and boiling temperature, related to the strength of intermolecular forces of attraction. In this experiment, you will study temperature changes caused by the evaporation of several liquids and ... 4 Chemistry with ...

Computer 9 Evaporation and Intermolecular Attractions copy

in terms of intermolecular forces. LEARNING OBJECTIVES: By the end of this experiment, the student should be able to demonstrate the following proficiencies: 1. Use a spreadsheet program for data manipulation, graphing, and regression analysis. 2. Describe the effects of changes in temperature on the vapor pressure of a pure substance. 3.

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