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Answer: 4 protons. Check: $4p + 5n = 9$ mass number Question: An atom has a mass number 11 and has 5 protons. How many neutrons does it have? Solution: mass number = number of protons (p) + number of neutrons (n). Mass Number = $p + n$ $11 = 5 + n - 5 ! 5 6 = n$ Answer: 6 neutrons. Check: $6n + 5p = 11$ mass number Now letâ€™s look at hydrogen.

Chapter 2: Atomic Concepts - High Marks

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